Synthesis of Spiroligomer-Containing Macrocycles

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S Supporting Information

ABSTRACT: We demonstrate the synthesis and characterization of the solution conformations of a collection of functionalized spiroligomer-based macrocycles. These macrocycles contain 14 independently controllable stereocenters and four independently controllable functional groups on a highly preorganized scaffold. These molecules are being developed to display complex, preorganized surfaces for binding proteins and to create enzyme-like active sites. In this work, we demonstrate the convergent synthetic approach to this new class of macrocycles and demonstrate that the conformational properties of these molecules can be changed by altering the configuration stereocenters within the backbone.

■ **INTRODUCTION**

Macrocyclic molecules have received significant attention because of their potential applications in molecular recognition, catalysis, material science, and pharmaceutical development. A number of preorganized macrocyclic scaffolds bearing welldefined structures have been developed, including calixarenes,^{1−6} cyclodextrin,^{7−11} resorcinarenes,^{12−17} conjugated aromatic systems,^{18−23} and Schiff base macrocycles,^{24−37} amo[ng](#page-9-0) [o](#page-10-0)thers.^{38–48} [Howe](#page-10-0)ver, site-specific[all](#page-10-0)y [i](#page-10-0)ncorporating multiple and diver[se](#page-10-0) f[unc](#page-10-0)tional groups, such as catalytic g[roups](#page-10-0) or protein bi[nding](#page-10-0) groups, within these scaffolds is often challenging due the inherent symmetry of many of the most studied macromolecules. In other cases, such as some Schiff base macrocycles, $25,34,35$ the cyclic scaffolds are assembled from monomers in one step, making it difficult to append multiple functional groups [in mo](#page-10-0)re than one specific positions. Proteins demonstrate that the presentation of diverse functional groups in preorganized three-dimensional constellations can give rise to tremendous capabilities, such as the catalysis in enzyme active sites and molecular recognition in protein binding sites. Therefore, new classes of synthetic macromolecular scaffolds that can present preorganized constellations of functional groups are highly desirable for the development of catalytic and molecular recognition applications.

Toward this goal, we have developed a collection of cyclic, stereochemically pure monomers called "bis-amino acids" that couple through pairs of amide bonds to form spiro- ladder oligomers named "spiroligomers". ⁴⁹ Along the way, a new amide bond forming reaction, called "acyl-transfer-coupling",⁵⁰

was developed and has been applied to the sequential formation of functionalized penta- and hexasubstituted diketopiperazines between pairs of functionalized bis-amino acids.⁵¹ Herein, we report a facile synthesis of spiroligomer-based macrocyclic scaffolds, whose structure and functional group [pre](#page-10-0)sentation is controlled by the sequence and stereochemistry of the component monomers. In addition, NMR studies showed that one macrocycle exists in a small number of preferred conformational isomers, and they exchange slowly on the NMR time scale at 298 K. The barrier between two of these isomers can be partially overcome at the slightly higher temperature of 333 K. Considering the 14 independently configurable stereocenters in the backbone and the four configurable functional groups on the macrocycles, the macromolecules presented here represent a few examples of a virtual library containing millions of different conformationally restrained, functionally diverse macrocycles with unexplored and diverse properties. The synthetic methodology we present here provides an effective path for the development of these molecules for future applications.

■ RESULTS AND DISCUSSION

Synthesis of Building Blocks: Preactivated pro4 Amino Acids 11−17. The general synthetic approach to preactivated, alkylated pro4 amino acids like 11−17 has been described in previous reports from our group.^{49,51} Taking

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Scheme 1. Synthesis of Preactivated Alkylated pro4 Amino Acids

Scheme 2. Synthesis of Spiroligomer Trimer

compound 11 as a representative example, free amine 2 was reductively alkylated overnight using 2-naphthylaldehyde and sodium cyanoborohydride in methanol at room temperature, with a quantitative yield as determined by LC−MS (Scheme 1).

The resulting N-alkylated amino acid 4 was treated with 6 equiv of 1-hydroxy-7-azabenzotriazole (HOAt) and 1 equiv of N,N′ diisopropylcarbodiimide (DIC) in dichloromethane/DMF (2:1) at room temperature for 2 h. Under these conditions, the resulting oxy-7-azabenzotriazole (−OAt) ester 11 will not polymerize with its own secondary amine because of the extremely hindered nature of the secondary amine on the quaternary carbon center.⁵¹ Instead, under dry atmosphere, the activated carboxylate species will remain in solution for many minutes, long enough to [p](#page-10-0)erform nucleophilic coupling with other added nucleophiles. By following this scheme, alkylated pro4 amino acids 5−10 were prepared with quantitative yields and preactivated as oxy-7-azabenzotriazole (−OAt) esters 12− 17 for assembly into functionalized spiroligomers.

Synthesis of Spiroligomer Trimers. Spiroligomer 26 was synthesized starting from intermediate 19 in seven steps with an overall 39% yield (Scheme 2, entry 1). Compound 19 was prepared by coupling 4-pentenoic acid with 1 equiv of N-Boc, $CO₂$ -tert-butyl-protect[ed hydroxy](#page-1-0)lproline 18 at room temperature, followed by removal of the protecting groups and precipitation of the amino acid 19 from diethyl ether/hexanes (1:1) solution. Amino acid 19 was then treated with 1.2 equiv of preactivated pro4 amino acid 11 as described above under basic conditions at room temperature. The coupling likely proceeds via multiple pathways including direct acylation of the prolyl amine of 19 and a new reaction mechanism that we have previously described called "acyl-transfer coupling".⁵⁰ The hexasubstituted diketopiperazine intermediate 22 was formed after addition of 2.0 equiv of DIC followed by remov[al o](#page-10-0)f the N-Boc and tert-butyl ester protecting groups using 95% TFA/ 2.5%TIPS/2.5% H_2O . The dimer intermediate 22 was then treated with another 1.2 equiv of preactivated pro4 amino acid 11 under basic conditions, followed by addition of DIC and

removal of the protecting groups. The resulting amino acid residue was protected with acetic anhydride to provide acylprotected spiroligomer trimer 26.

The synthesis of these spiroligomer trimers is extremely versatile. Every stereocenter can be controlled and every functional group can be substituted. Starting from 20 and 21, trimer 27 and 28 with shorter alkene linkers were obtained by following the identical synthetic route as that for trimer 26 (Scheme 2, entry 2 and 3). Trimer 29 was prepared with a 33% overall yield by following the same protocol as trimer 26 except [alkylated](#page-1-0) pro4 monomer 12 was used in place of 11 (Scheme 2, entry 4). Different side-chain groups $(R_1 \text{ and } R_2)$ can be incorporated by sequential addition of alkylated pro4 monomers (Scheme 2, entries 5−8) bearing different [functional](#page-1-0) groups derived from appropriate aldehydes. The shape of the trimer, as w[ell as the](#page-1-0) relative positions of side chain groups in three-dimensional space, can be controlled by utilizing pro4 monomers with different stereochemistry (Scheme 2, entries 5 and 6).

Macrocycle Formation through a [Metathes](#page-1-0)is Reaction. Trimers 26−28 were coupled with 0.5 equiv of either ethylenediamine or diaminocyclohexane to provide dialkenes 34−37 (Scheme 3). The metathesis ring closure ring of 34 was not achieved using Grubbs I, II or Hoveyda−Grubbs I catalysts either at room temperature or under dichloromethane reflux. With 50% Hoveyda−Grubbs II catalyst in dichloromethane at room temperature, the metathesis reaction of 34 proceeded smoothly to afford 38 in 72% yield. Although both E and Z configurations may be formed as products, only one peak was observed in the C18 reversed-phase LC−MS chromatograph. The E/Z configuration of the alkene could not be determined conclusively; therefore, the next step was to reduce it to the alkane for NMR analysis. Similarly, macrocycles 39 and 40 were

Scheme 4. Synthesis of Non- C_2 -Symmetric Tetrasubstituted Macrocycles 46 and 47

formed under the same conditions from the corresponding precursors 35 and 36. The metathesis reaction of substrate 37 failed with decomposition, as observed by LC−MS.

Synthesis of Tetrasubstituted Macrocycles 46 and 47. With an olefin metathesis pathway to the macrocyles developed, a modification of the synthesis was carried out to produce non- C_2 -symmetric tetrasubstitued macrocycles. Slow addition of 1-[dis(dimethylamino)methylene]-1H-1,2,3 triazolo[4,5-b]pyridinium 3-oxide hexafluorophosphate (HATU) activated trimer 32 into 6 equiv of diaminocyclohexane in DMF resulted in only one of the amines of the diamine being acylated by the spiroligomer segment, with the other primary amine free (Scheme 2, entries 7 and 8). The primary amine 33 was coupled with trimer 30 or 31 to form the dialkene intermediate 42 or 43, which was then subjected sequentially to metat[hesis](#page-1-0) [reacti](#page-1-0)on and hydrogenation using $H₂$ catalyzed by palladium on carbon to produce macrocycles 46 and 47 (Scheme 4). Since trimers were incorporated sequentially into the scaffold, this synthesis allows them to be different from each other and enables the synthesis of completely asymmetric macrocycles.

Conformational Analysis of Macrocyclic 46 and 47. A series of NMR experiments including ¹H spectra, COSY, ROESY, as well as the ${}^{1}H, {}^{13}C$ heteronuclear experiments HMQC and HMBC were performed to determine if the macrocyclic spiroligomers 46 and 47 displayed well-defined structures or if they were dynamic in benzene- d_6 . Nearly all the 1 H and 13 C signals were assigned as shown in the Supporting Information. In addition, computational conformational searching using the AMBER94 force field using the [Molecular](http://pubs.acs.org/doi/suppl/10.1021/acs.joc.5b01109/suppl_file/jo5b01109_si_001.pdf)

Operating Environment (MOE) was performed to identify the lowest energy conformers of these two molecules.

Macrocycle 46 (Figure 1A) is composed of four pro4 (2S,4S) monomers having three aromatic (M, N, and P) and one isopropyl (Q) g[roups as](#page-4-0) side chains. The structures of individual functionalized spiroligomer segments and the presentation of functional groups M and N (as well as P and Q) are well-defined by virtue of the ladder structure of spiroligomers. In these experiments, we were interested to observe if the spiroligomer segments have any predisposition to align with each other because of the cross-links across the two ends of the segments. The conformational searches of macrocycle 46 suggest that they are quite dynamic. This can be seen in two conformers extracted from a molecular dynamics run followed by energy minimization of compound 46 (Figure 1C). The side chains of segment MN are dynamic and sample multiple rotamers. In the dynamics simulations, th[e MN](#page-4-0) [se](#page-4-0)gment moves quite freely relative to the PQ segment because the hydrocarbon linker formed using the metathesis reaction is relatively long and flexible. The 4-bromobenzyl group (M) and 4-nitrobenzyl group (P) have a strong tendency to sit on opposite faces of the macrocycle ring with a distance of about 10.6 Å. In the ROESY NMR experiment, cross-peaks between the protons on the side chains and the spiroligomer segments that they are directly attached to were observed and were consistent with the ladder-molecule structure of the spiroligomer segments (Figure 1B); however, no ROESY crosspeaks were observed between hydrogens from any atoms on one spiroligomer se[gment to i](#page-4-0)ts neighboring segment. The lack of observable ROESY correlations between side-chains can be

Figure 1. (A) Structure of macrocycle 46; (B) ROESY correlations observed in NMR experiment are shown using blue, red, and magenta arrows; (C) two low energy conformers of 46 after energy minimization (ring M was treated as a benzyl group because Br atom is not recognized in AMBER94 force field).

understood in terms of the dynamic nature of macrocycles 46 as suggested by the molecular dynamics simulations.

Macrocycle 47 has the identical bonded structure as that of 46 except one pro4 (2S4R) monomer was incorporated rather than a pro4 (2S4S) monomer in one segment. This has a large consequence to the modeled structure of 47 relative to 46 in that it flips the side chains M and N to the same face of the macrocycle as P and Q. Thus, macrocycle 47 is a diastereomer of 46 that differs in only one of the 14 stereocenters, and therefore, 46 and 47 are two diastereomers of a set of 16384 total stereoisomers. Molecular dynamics simulations of 47 indicate that 47 is also highly dynamic but involves very different side-chain presentations than 46.⁵²

An unanticipated conformational isomerism was observed in the COSY NMR experiment of compound [4](#page-10-0)7 in benzene- d_6 as shown in Figure 2. At 298 K, the coupling between BCHG and BCHB1 (or BCHB2) was split into two separate cross-peaks (Figure 2A), both of which coupled with one carbon resonance (Figure 2B). At 333 K, the coupling between these two protons coalesced into one single cross peak as shown in Figure 2D. Although a slightly different chemical shift was observed due to the increased temperature (Figure $2C$), it suggests that compound 47 exists in a restricted set of conformations that exchange slowly on the NMR time-scale at 298 K.

The restricted conformational flexibility of 47 and the slowly exchanging conformational isomerism is also visible in the aromatic area in the COSY spectrum as shown in Figure 3. At 298 K, unlike the single cross-peak shown in macrocycle 46 (Figure 3A), the coupling between HM3 and H[M2 on the](#page-5-0) p-

Figure 2. Observation of slow exchange on the NMR time scale between at least two conformations in 47 at 298 K that coalesce at 333 K- protons in aliphatic carbon (labeling rules are discussed in Supporting Information).

[bromobenzyl group](http://pubs.acs.org/doi/suppl/10.1021/acs.joc.5b01109/suppl_file/jo5b01109_si_001.pdf) (ring M) was split into three separate cross-peaks (Figure 3B). At 333 K, this coupling partially coalesced into two cross-peaks (Figure 3C). It is noteworthy that the *p*-br[omobenzy](#page-5-0)l group (side chain M) is very close to the stereocenter that is differe[nt betwee](#page-5-0)n 46 and 47. The chemical shift of protons in the 4-methoxybenzyl group (ring N), which was on the same spiroligomer fused ring scaffold, moved upfield after flipping the stereocenter, whereas the chemical shift of protons in the 4-nitrobenzyl group (ring Q) are unchanged (Figure 3D). These observations suggest that compound 47 has a restricted set of conformational isomers, and they exchan[ge slowly](#page-5-0) on the NMR time scale at 298 K. The barrier between two of these isomers can be partially overcome at a slightly higher temperature (333 K). This demonstrates that changing just one stereocenter out of the 14 stereocenters of the backbone can lead to different conformational properties

Figure 3. Observation of slow exchange in the ¹H COSY on the NMR time-scale between well-defined conformations in macrocycle 47 at 298 K, some of which coalesce at 333 K; the protons are in aromatic rings.

of the macrocycles. In the future, we plan to incorporate shorter and more constrained linkers as well as careful design of the stereocenters of the spiroligomer segments to reduce the dynamics of the macrocycles and bias the structure to one welldefined three-dimensional structure that presents the functional groups in constellations that enable them to bind protein surfaces 53 and accelerate chemical reactions. $54-56$

■ C[ON](#page-10-0)CLUSION

In summary, a synthetic protocol of macrocyclic spiroligomers using cross-olefin metathesis coupling was developed, and a series of tetrafunctionalized macromolecules were synthesized. The NMR study of compounds 46 and 47 indicated that they are conformationally dynamic, and this is consistent with molecular dynamics calculations using the AMBER94 force field. Compound 47 exhibits a limited number of conformational isomers as demonstrated by well-defined cross peaks in the correlations isomers, and they exchange slowly on the NMR time scale at 298 K. The barrier between two of these isomers can be partially overcome at the slightly higher temperature (333 K). The characters of the similar macrocycles with shorter linkers are currently under investigation. This synthetic methodology also provides a path to create more spiroligomer based macromolecules to create molecular surfaces and pockets that will be used to develop applications in supramolecular molecular recognition and catalysis.

EXPERIMENTAL SECTION

General Synthetic Procedure of Alkylated pro4 Amino Acid 4−10. The synthesis of pro4 amino acids 2 and 3 has been previously described in the literature.⁴⁹ A general synthetic procedure-reductive alkylation of pro4 amino acids is described as follows: To a solution of (3S,5S)-3-amino-1,5-bis(t[ert](#page-10-0)-butoxycarbonyl)pyrrolidine-3-carboxylic acid 2 (300 mg to 2 g scale, 1.0 equiv) in methanol (10 mL per mmol of 2), an aldehyde chosen from the table in Scheme 1 (1.2 equiv) was added and stirred for 30 min at room temperature. Sodium cyanoborohydride (1.5 equiv) was then added and stirred for 20 h at room temperature until the reaction was [complete, a](#page-1-0)s monitored by LC−MS. The reaction mixture was concentrated under vacuum and the resulting residue was dissolved in 50 mL of H_2O . The pH was adjusted to 7 by dropwise addition of 2 N HCl. Crude alkylated pro4 amino acid $(4-10)$ was obtained by vacuum filtration with quantitative yield. (Note: this reaction is a peak-to-peak reaction as determined by LC−MS, and the products formed were used directly without further purification in the next steps, so the purified isolated yields were not calculated. For characterization purposes, alkylated amino acids were purified by reversed- phase C_{18} high-performance liquid chromatography using a 5−95% H2O/acetonitrile gradient).

(3S,5S)-1,5-Bis(tert-butoxycarbonyl)-3-((naphthalen-1-ylmethyl) amino)pyrrolidine-3-carboxylic acid, 4: ¹H NMR (500 MHz, DMSO- d_6 , 350 K) δ 8.20 (d, J = 8.0 Hz, 1H), 7.88 (dd, J = 7.8, 1.5 Hz, 1H), 7.78 (d, J = 8.2 Hz, 1H), 7.54–7.46 (m, 3H), 7.41 (dd, J = 8.0, 7.2 Hz, 1H), 4.16 (dd, J = 8.5, 6.7 Hz, 1H), 4.08 (q, J = 12.9 Hz, 2H), 3.93 (d, $J = 10.6$ Hz, 1H), 3.33 (d, $J = 10.6$ Hz, 1H), 2.74 (s, 1H), 2.00 (dd, J = 12.7, 6.6 Hz, 1H), 1.39 (s, 9H), 1.36 (s, 9H); ¹³C NMR (126 MHz, DMSO- d_6 , 350 K) δ 175.3, 171.5, 153.6, 136.5, 133.8, 132.0, 128.7, 127.7, 126.5, 126.2, 125.8, 125.7, 124.3, 80.6, 79.2, 59.6, 55.2, 46.7, 28.5, 28.3, 28.1; HRMS-ESI m/z calcd for $C_{26}H_{34}N_2O_6$ (M $+ H$ ⁺ 471.2490, found 471.2488.

(3S,5S)-3-((4-((Benzyloxy)carbonyl)benzyl)amino)-1,5-bis(tertbutoxycarbonyl)pyrrolidine-3-carboxylic acid, 5: ¹H NMR (500 MHz, DMSO- d_6 , rt, rotamers observed) δ 8.05 (s, 1H), 7.98–7.85 (m, 2H), 7.55−7.41 (m, 4H), 7.40−7.30 (m, 3H), 5.33 (s, rotamers, 2H), 4.15 (t, rotamers, J = 7.3 Hz, 1H), 3.98−3.56 (m, 3H), 3.33 (d,

rotamers, $J = 10.6$ Hz, 1H), 2.67 (dd, rotamers, $J = 11.9$, 9.0 Hz, 1H), 2.04 (dd, rotamers, J = 12.0, 6.1 Hz, 1H), 1.40 (s, 9H), 1.34 (s, 9H); ¹³C NMR (126 MHz, DMSO-d₆, rt, rotamers observed) δ 174.8, 171.4, 171.1, 167.6, 166.0, 165.6, 153.8, 153.4, 149.0, 146.2, 146.0, 137.7, 136.7, 136.4, 132.9, 129.9, 129.6, 129.5, 128.94, 128.85, 128.7, 128.6, 128.5, 128.3, 126.8, 81.5, 80.8, 80.6, 79.7, 79.3, 68.2, 67.2, 66.9, 66.5, 62.9, 59.3, 54.7, 48.6, 38.8, 38.1, 28.5, 28.4, 28.3, 28.1; HRMS-ESI m/z calcd for $C_{30}H_{38}N_2O_8$ $(M + H)^+$ 555.2701, found 555.2699.

(3S,5S)-1,5-Bis(tert-butoxycarbonyl)-3-((4-methoxybenzyl) amino)pyrrolidine-3-carboxylic acid, $6:$ $^1\mathrm{H}$ NMR (500 MHz, DMSO-d6, 350 K) δ 7.24−7.21 (m, 2H), 6.87−6.84 (m, 2H), 4.14 $(dd, J = 8.6, 6.4 Hz, 1H), 3.84 (d, J = 10.9 Hz, 1H), 3.74 (s, 3H), 3.60$ $(dd, J = 27.7, 12.6 Hz, 2H), 3.31 (d, J = 10.9 Hz, 1H), 2.66 (dd, J =$ 10.8, 9.8 Hz, 1H), 1.99 (dd, J = 12.9, 6.4 Hz, 1H), 1.41 (s, 9H), 1.39 (s, 9H); ¹³C NMR (126 MHz, DMSO- d_6 , 350 K) δ 174.2, 171.3, 159.0, 153.5, 132.0, 129.7, 114.1, 80.9, 79.4, 67.2, 59.3, 55.6, 54.7, 48.4, 38.7, 28.5, 28.1; HRMS-ESI m/z calcd for $C_{23}H_{34}N_2O_7$ $(M + H)^+$ 451.2439, found 451.2438.

(3S,5S)-3-((4-Bromobenzyl)amino)-1,5-bis(tert-butoxycarbonyl) pyrrolidine-3-carboxylic acid, 7: $^1\rm H$ NMR (500 MHz, DMSO- d_6 , 350 K) δ 7.47–7.43 (m, 2H), 7.29–7.26 (m, 2H), 4.13 (dd, J = 8.7, 6.2 Hz, 1H), 3.82 (d, J = 10.8 Hz, 1H), 3.61 (dd, J = 31.8, 13.5 Hz, 2H), 3.28 $(d, J = 10.9 \text{ Hz}, 1\text{H}), 2.70-2.61 \text{ (m, 1H)}, 1.98 \text{ (dd, } J = 13.0, 6.1 \text{ Hz},$ 1H), 1.39 (s, 9H), 1.38 (s, 9H); ¹³C NMR (126 MHz, DMSO- d_6 , rt, rotamers observed) δ 174.3, 171.4, 171.0, 165.1, 153.8, 153.4, 139.5, 131.3, 130.8, 81.0, 80.8, 79.5, 67.8, 66.9, 59.2, 54.6, 48.1, 38.7, 37.9, 28.5, 28.4, 28.1; HRMS-ESI m/z calcd for $C_{22}H_{31}BrN_2O_6 (M + H)^+$ 499.1438, found 499.1443.

(3R,5S)-3-((4-Bromobenzyl)amino)-1,5-bis(tert-butoxycarbonyl) pyrrolidine-3-carboxylic acid, \boldsymbol{s} : ^1H NMR (500 MHz, DMSO- d_6 , 350 K) δ 7.45 (d, J = 8.3 Hz, 2H), 7.26 (d, J = 8.3 Hz, 2H), 4.16 (t, J = 8.1 Hz, 1H), 3.69 (d, J = 13.9 Hz, 1H), 3.59 (d, J = 14.0 Hz, 1H), 3.55 (broad, 2H), 2.39−2.30 (m, 1H), 2.23−2.14 (m, 1H), 1.42 (s, 9H), 1.37 (s, 9H); ¹³C NMR (101 MHz, DMSO- d_6 , rt, rotamers observed) δ 177.4, 176.1, 174.9, 173.3, 173.2, 173.1, 171.7, 171.3, 168.2, 166.6, 154.3, 153.8, 153.5, 153.3, 151.3, 140.2, 139.9, 131.3, 130.6, 120.1, 99.4, 80.9, 80.7, 79.3, 68.3, 67.5, 61.4, 59.3, 59.1, 28.3, 28.0; HRMS-ESI m/z calcd for $C_{22}H_{31}BrN_2O_6$ $(M + H)^+$ 499.1438, found 499.1436.

(3S,5S)-1,5-Bis(tert-butoxycarbonyl)-3-(isobutylamino) pyrrolidine-3-carboxylic acid, **9**: 1 H NMR (500 MHz, DMSO- d_{ω} 350 K) δ 4.12 (t, J = 7.3 Hz, 1H), 3.81 (d, J = 10.6 Hz, 1H), 3.26 (d, J = 10.6 Hz, 1H), 2.63 (broad, 1H), 2.34 (dd, J = 10.9, 6.4 Hz, 1H), 2.29− 2.21 (m, 1H), 1.91 (dd, J = 12.3, 6.2 Hz, 1H), 1.67−1.56 (m, 1H), 1.44 (s, 9H), 1.39 (s, 9H), 0.88 (dd, J = 6.2, 2.9 Hz, 6H); 13C NMR (126 MHz, DMSO- d_6 , 350 K, aliphatic carbon signals overlap) δ 174.0, 171.5, 171.2, 80.8, 79.3, 59.4, 54.7, 52.7, 28.5, 28.1, 20.7; HRMS-ESI m/z calcd for $C_{19}H_{34}N_2O_6$ $(M + Na)^+$ 409.2309, found 409.2316.

(3S,5S)-1,5-Bis(tert-butoxycarbonyl)-3-((4-nitrobenzyl)amino) pyrrolidine-3-carboxylic acid, 10: ^1H NMR (500 MHz, DMSO- d_{6} , 350 K) δ 8.14–8.10 (m, 2H), 7.63–7.59 (m, 2H), 4.15 (dd, J = 8.7, 6.0 Hz, 1H), 3.87–3.73 (m, 3H), 3.31 (d, J = 10.9 Hz, 1H), 2.68 (dd, J $= 10.4, 9.5$ Hz, 1H), 2.01 (dd, J = 13.0, 6.0 Hz, 1H), 1.40 (s, 9H), 1.38 (s, 9H); ¹³C NMR (101 MHz, DMSO- d_6 , rt, rotamers observed) δ 174.6, 171.1, 170.7, 153.6, 153.3, 149.0, 148.9, 146.7, 129.3, 123.4, 80.8, 80.7, 79.33, 79.28, 67.7, 66.8, 59.0, 54.7, 54.6, 48.0, 38.8, 37.9, 28.4, 28.2, 27.9; HRMS-ESI m/z calcd for $C_{22}H_{31}N_3O_8$ $(M + H)^+$ 466.2184, found 466.2188.

(2S,4R)-4-(Pent-4-enoyloxy)pyrrolidine-2-carboxylic Acid, 19. At room temperature, alcohol 18 (synthesized from 4-hydroxyproline, $57,58$ 1.5 g, 5.2 mmol, 1.05 equiv), 4-pentenoic acid (500 mg, 5.0) mmol, 1.0 equiv), and DMAP (61 mg, 0.5 mmol, 0.1 equiv) were diss[olved](#page-10-0) in 50 mL of DCM. To this solution was added DIC (862 μ L, 5.5 mmol, 1.1 equiv) dropwise. The reaction was stirred at room temperature overnight until completion as detected by TLC with I_2 stain. The reaction was then quenched with 50 mL of satd $NH₄Cl$ (aq), and the product was extracted with DCM $(3 \times 100 \text{ mL})$. The combined organic layers were washed subsequently with satd $NH₄Cl$ (aq), satd NaHCO₃ (aq), and brine and dried over anhydrous Na₂SO₄. After removal of the solvent, the crude product was purified through silica gel chromatography with 0−15% EtOAc/hexanes as eluting solvent to give a white solid intermediate (1.7 g, 4.6 mmol, 92% yield). The intermediate (1.5 g, 4.1 mmol) was then treated with 50 mL of 95% TFA/2.5% TIPS/2.5% H_2O at room temperature for 2 h. Removal of the solvent produced a light yellow oil, which was then added dropwise into 100 mL of ether/hexanes (1:1) and stirred for 30 min. The white precipitate formed was then filtered and dried under high vacuum to provide the white solid powder 19 as a TFA salt (1.02 g, 76% yield): ¹H NMR (500 MHz, DMSO- d_5 , rt) δ 5.80 (ddt, J = 16.7, 10.2, 6.4 Hz, 1H), 5.26 (t, J = 4.8 Hz, 1H), 5.04 (ddd, J = 17.2, 3.4, 1.6 Hz, 1H), 4.98 (ddd, $J = 10.3, 3.1, 1.3$ Hz, 1H), 4.17 (dd, $J =$ 10.2, 7.8 Hz, 1H), 3.50 (dd, J = 13.0, 4.8 Hz, 1H), 3.22 (d, J = 13.0 Hz, 1H), 2.43−2.35 (m, 2H), 2.32−2.23 (m, 3H), 2.23−2.16 (m, 1H); 13C NMR (126 MHz, DMSO- d_6 , rt) δ 172.2, 170.1, 137.4, 116.0, 73.3, 59.0, 50.8, 35.1, 33.1, 28.6; HRMS-ESI m/z HRMS-ESI m/z calcd for $C_{10}H_{15}NO_4$ $(M + H)^+$ 214.1074, found 214.1080.

(2S,4R)-4-(But-3-enoyloxy)pyrrolidine-2-carboxylic Acid, 20. Compound 20 was prepared by following the identical synthesis procedure as compound 19 except 4-pentenoic acid was changed to 3 butenoic acid: ¹H NMR (500 MHz, DMSO- d_6 , rt) δ 10.05 (s, 1H), 9.01 (s, 1H), 5.85 (ddt, J = 17.1, 10.2, 6.9 Hz, 1H), 5.29 (t, J = 4.8 Hz, 1H), 5.17−5.09 (m, 2H), 4.45 (t, J = 8.6 Hz, 1H), 3.56 (d, J = 10.2 Hz, 1H), 3.32 (d, J = 12.9 Hz, 1H), 3.11 (ddt, J = 6.9, 4.0, 1.4 Hz, 2H), 2.38 (ddt, J = 8.0, 3.3, 1.4 Hz, 1H), 2.34–2.26 (m, 1H); ¹³C NMR $(126 \text{ MHz}, \text{DMSO-}d_6, \text{ rt}) \delta 170.6, 170.1, 130.7, 118.8, 72.9, 58.1, 50.9,$ 38.5, 34.4; HRMS-ESI m/z HRMS-ESI m/z calcd for $C_9H_{13}NO_4$ (M + Na)⁺ 222.0737, found 222.0738. Compound 21 has been reported in the previous literature.^{59,60}

General Synthetic Procedure of Spiroligomer Trimer. A representative synthes[is of](#page-10-0) trimer 26 is described as below: At room temperature, pro4 amino acid 4 (940 mg, 2.0 mmol, 1.0 equiv) and HOAt (1.6 g, 12.0 mmol, 6.0 equiv) were dissolved in 30 mL of DMF/ DCM (2:1) in a flame-dried 100 mL round-bottom flask. To this solution was added DIC (344 μ L, 2.2 mmol, 1.1 equiv) dropwise, and the reaction was stirred at room temperature for 2 h. To the reaction with intermediate 11 formed in situ were then added the premixed 19 (556 mg, 1.7 mmol, 0.85 equiv) and DIPEA (883 μ L, 5.1 mmol, 2.55 equiv) in 10 mL of DMF in one portion. The reaction was stirred for 15 h, followed by the addition of DIC (626 μ L, 4.0 mmol, 2.0 equiv), and it stirred for an additional 2 h until the DKP closure was complete as detected by LC−MS. The reaction mixture was then quenched by the addition of satd NH₄Cl (aq) (40 mL) and extracted with EtOAc (100 mL + 2×50 mL). The combined organic portion was washed sequentially with $3\times$ satd NH₄Cl (aq), $3\times$ satd NaHCO₃ (aq), and $2\times$ brine and dried over anhydrous $Na₂SO₄$. After filtration and removal of solvent under vacuum, the yellow solid residue was treated with 50 mL of 95% TFA/2.5% TIPS/2.5% H_2O at room temperature for 2 h. Removal of the solvent under vacuum produced a yellow oil crude product, which was then added dropwise into 200 mL of ether/ hexanes (1:1) and stirred for 30 min. The precipitate formed was then filtered and dried under high vacuum to give the light yellow solid dimer 22 as a TFA salt (1.49 g). The product was used in the next step without further purification: HRMS-ESI m/z calcd for $C_{27}H_{30}N_3O_6^{\frac{1}{2}}$ $(M + H)^+$ 492.2129, found 492.2121.

At room temperature, amino acid 4 (553 mg, 1.18 mmol, 1.0 equiv) and HOAt (964 mg, 7.08 mmol, 6.0 equiv) were dissolved in 18 mL of DMF/DCM (2:1) in a flame-dried 100 mL round-bottom flask. To this solution was added DIC (203 μ L, 1.29 mmol, 1.1 equiv) dropwise, and the reaction was stirred at room temperature for 2 h. To the reaction were then added the premixed crude dimer 22 (606 mg, theoretically 1.0 mmol, 0.85 equiv) and DIPEA (520 μ L, 3.0 mmol, 2.55 equiv) in 6 mL DMF in one portion. The reaction was stirred for 15 h followed by the addition of DIC (369 μ L, 2.35 mmol, 2.0 equiv) and stirred for an additional 2 h until the DKP closure was complete as indicated by LC−MS. The reaction mixture was then quenched by addition of satd NH₄Cl (aq) (40 mL), and product was extracted with EtOAc (100 mL + 2 \times 50 mL). The combined organic portion was washed sequentially with $3\times$ satd NH₄Cl (aq), $3\times$ satd NaHCO₃ (aq), and $2\times$ brine and dried over anhydrous Na₂SO₄. After filtration and removal of solvent under vacuum, the yellow solid residue was treated with 50 mL of 95% TFA/2.5% TIPS/2.5% $H₂O$ at room temperature for 2 h. Removal of the solvent under vacuum produced a crude product as a dark yellow oil, which was then added dropwise into 200 mL of ether/hexanes (1:1) and stirred for 30 min. The precipitate formed was then filtered and dried under high vacuum to give the crude brown solid trimer as a TFA salt (1.03 g). The crude product was used in the next acyl protection step without further purification.

The crude product obtained above (883 mg, theoretically 1 mmol, 1.0 equiv) was dissolved in 20 mL of THF, followed by the addition of DIPEA (520 μ L, 3.0 mmol, 3.0 equiv) and Ac₂O (142 μ L, 1.5 mmol, 1.5 equiv). The reaction was stirred overnight until completion as indicated by LC−MS. The reaction mixture was then quenched by addition of satd NH₄Cl (aq) (40 mL), and the product was extracted with EtOAc (100 mL + 2×50 mL). The combined organic portion was washed sequentially with $3\times$ satd NH₄Cl (aq) and $2\times$ brine and dried over anhydrous Na₂SO₄. After filtration and removal of solvent under vacuum, the crude product 26 was obtained as a brown solid. The pure trimer 26 (190 mg) was obtained after purification by reversed-phase C₁₈ column using 5−95% H₂O/acetonitrile gradient as the eluting solvent with an overall 39% isolated yield from the starting building block 19.

(3S,5S,7′S,7″R,8′aS,8″aS)-1-Acetyl-2′,2″-bis(naphthalene-1-ylmethyl)-1′,1″,4′,4″ tetraoxo-7″-(pent-4-enoyloxy)dodecahydro-1″H-dispiro[pyrrolidine-3,3′:7′,3″-bis(pyrrolo[1,2-a]piperazine)]-5- $\frac{1}{100}$ carboxylic acid, 26: $\frac{1}{1}$ NMR (500 MHz, DMSO- d_6 , 350 K, rotamer observed) δ 8.12 (d, J = 8.1 Hz, 2H), 8.02–7.96 (m, 1H), 7.96–7.90 $(m, 1H)$, 7.85 $(d, J = 8.2 \text{ Hz}, 1H)$, 7.79 $(d, J = 8.2 \text{ Hz}, 1H)$, 7.63–7.57 (m, 2H), 7.57−7.52 (m, 2H), 7.46 (dd, J = 7.9, 7.5 Hz, 1H), 7.38 (t, J $= 7.5$ Hz, 1H), 7.14 (d, J = 6.9 Hz, 1H), 6.99 (s, 1H), 5.86 (ddt, J = 16.7, 10.3, 6.4 Hz, 1H), 5.50 (d, J = 17.5 Hz, 1H), 5.37 (t, J = 4.8 Hz, 1H), 5.31−5.24 (m, 2H), 5.08 (dq, J = 17.2, 1.7 Hz, 1H), 5.01 (ddd, J $= 10.3, 3.1, 1.4$ Hz, 1H), 4.94 (d, J = 17.5 Hz, 1H), 4.82 (dd, J = 10.1, 7.3 Hz, 1H), 4.59−4.48 (m, 1H), 4.15 (dd, J = 13.5, 5.4 Hz, 1H), 4.03 $(s, J = 7.6 \text{ Hz}, 1\text{H})$, 3.99 (d, J = 12.4 Hz, 1H), 3.76 (dd, J = 12.8, 3.9 Hz, 1H), 3.55 (d, $J = 13.6$ Hz, 1H), 3.34 (dd, $J = 35.3$, 12.7 Hz, 1H), 2.96 (dd, J = 13.2, 6.7 Hz, 1H), 2.70 (dd, J = 21.0, 9.3 Hz, 1H), 2.60 $(dd, J = 13.4, 9.8$ Hz, 1H), 2.47 $(d, J = 7.1$ Hz, 2H), 2.44–2.21 (m, 6H), 1.80 (s, J = 12.6 Hz, 3H); ¹³C NMR (126 MHz, DMSO- d_6 , 350 K, rotamer observed) δ 173.0, 172.3, 169.2, 168.8, 168.6, 166.2, 165.7, 137.5, 134.3, 133.8, 133.7, 133.3, 130.5, 129.1, 128.9, 127.8, 127.3, 126.7, 126.6, 126.4, 126.2, 125.9, 125.8, 123.6, 123.3, 122.2, 115.8, 71.9, 71.1, 68.2, 67.6, 62.2, 58.8, 57.0, 56.8, 52.7, 51.4, 51.2, 50.6, 44.5, 43.9, 42.7, 38.2, 36.1, 34.2, 33.5, 28.7, 22.3; LC−MS analysis at 274 nm ($\rm{C_{18}}$ reversed-phase, 40 min, 5−100% $\rm{H_2O}/$ acetonitrile with 0.1% formic acid) $t_R = 21.8$ min; HRMS-ESI m/z calcd for $C_{46}H_{45}N_5O_9$ (M + H)+ 812.3290, found 812.3290.

(3S,5S,7′S,7″R,8′aS,8″aS)-1-Acetyl-7″-(but-3-enoyloxy)-2′,2″-bis- (naphthalene-1-ylmethyl)-1′,1″,4′,4″tetraoxododecahydro-1″Hdispiro[pyrrolidine-3,3′:7′,3″-bis(pyrrolo[1,2-a]piperazine)]-5-carboxylic Acid, Trimer 27. Starting from amino acid 20 (356 mg, 1.14 mmol), trimer 27 was synthesized through dimer intermediate 23 by following the identical synthetic procedure as that of trimer 26 with an overall 67% isolated yield: 1 H NMR (500 MHz, DMSO- d_{6} , 350 K) δ 8.16−8.03 (m, 2H), 7.98 (t, J = 8.7 Hz, 2H), 7.89−7.81 (m, 2H), 7.62−7.55 (m, 4H), 7.50−7.39 (m, 2H), 7.23−7.02 (m, 2H), 5.94 $(ddt, J = 17.0, 10.3, 6.8 Hz, 1H), 5.52 (d, J = 17.5 Hz, 1H), 5.40 (t, J =$ 4.8 Hz, 1H), 5.36 (d, J = 17.4 Hz, 1H), 5.22 (dd, J = 17.2, 1.6 Hz, 1H), 5.18 (dd, J = 10.2, 1.4 Hz, 1H), 5.03 (broad, 1H), 4.94 (d, J = 17.3 Hz, 1H), 4.83 (dd, J = 10.1, 7.4 Hz, 1H), 4.64 (dd, J = 10.2, 6.9 Hz, 1H), 4.41−4.28 (m, 1H), 4.18 (dd, J = 13.5, 5.4 Hz, 1H), 4.15−4.08 (m, 1H), 4.01 (d, J = 12.5 Hz, 1H), 3.89 (d, J = 12.2 Hz, 1H), 3.78 (d, J = 12.5 Hz, 1H), 3.56 (d, J = 13.6 Hz, 1H), 3.48 (dd, J = 15.8, 13.1 Hz, 1H), 3.19 (dd, J = 6.7, 0.8 Hz, 2H), 3.18−3.11 (m, 1H), 2.99 (dd, J = 13.2, 6.7 Hz, 1H), 2.78 (dd, J = 12.8, 6.5 Hz, 2H), 2.49–2.35 (m, 4H); ¹³C NMR (126 MHz, DMSO-d₆, 350 K, 5 aromatic and 3 aliphatic carbon signals were missing probably because of overlap) δ 170.80, 170.79, 168.7, 168.6, 168.4, 165.6, 165.1, 133.8, 133.2, 133.0, 131.0, 130.4, 130.3, 129.1, 127.7, 126.8, 126.6, 126.3, 125.9, 123.3, 122.9, 122.7, 122.2, 118.8, 71.4, 68.3, 67.5, 56.9, 56.7, 52.7, 52.2, 50.6, 49.8, 43.9, 38.8, 38.1, 36.0; LC−MS analysis at 274 nm (C₁₈ reversed-phase,

40 min, 5−100% H₂O/acetonitrile with 0.1% formic acid) $t_R = 21.0$ min; HRMS-ESI m/z calcd for $C_{45}H_{43}N_5O_9$ $(M + H)^+$ 798.3134, found 798.3167.

(3S,5S,7′R,8a′S)-7′-(But-3-enoyloxy)-2′-(naphthalen-1-ylmethyl)- 1′,4′-dioxohexahydro-1′H-spiro[pyrrolidine-3,3′-pyrrolo[1,2-a] pyrazine]-5-carboxylic acid, dimer intermediate 23: HRMS-ESI m/z calcd for $C_{26}H_{27}N_3O_6$ $(M + H)^+$ 478.1973, found 478.1970.

(3S,5S,7′S,7″R,8′aS,8″aS)-1-Acetyl-2′,2″-bis(naphthalene-1-ylmethyl)-1′,1″,4′,4″tetraoxo-7″-(prop-2-enoyloxy)dodecahydro-1″H-dispiro[pyrrolidine-3,3′:7′,3″-bis(pyrrolo[1,2-a]piperazine)]-5 carboxylic Acid, Trimer 28. Starting from amino acid 21 (160 mg, 0.53 mmol), trimer 28 was synthesized through dimer intermediate 24 by following the identical synthetic procedure as that of trimer 26 with an overall 19% isolated yield: ${}^{1}\text{H}$ NMR (500 MHz, DMSO- d_{6} , 350 K) δ 8.14−8.09 (m, 2H), 7.99−7.96 (m, 1H), 7.94−7.92 (m, 1H), 7.85 (d, J = 8.3 Hz, 1H), 7.79 (d, J = 8.2 Hz, 1H), 7.62–7.58 (m, 2H), 7.57−7.50 (m, 4H), 7.47 (dd, J = 8.1, 7.3 Hz, 1H), 7.42−7.35 (m, 1H), 7.15 (d, J = 7.1 Hz, 1H), 6.43 (dd, J = 17.3, 1.4 Hz, 1H), 6.21 $(dd, J = 17.3, 10.5 Hz, 1H), 5.98 (dd, J = 10.5, 1.4 Hz, 1H), 5.50 (d, J)$ = 17.5 Hz, 1H), 5.45 (td, J = 4.2, 1.1 Hz, 1H), 5.36−5.26 (m, 2H), 5.26−5.13 (m, 2H), 4.95 (d, J = 17.4 Hz, 1H), 4.87 (dd, J = 10.0, 7.5 Hz, 1H), 4.53 (dd, $J = 10.7$, 6.7 Hz, 1H), 4.20 (dd, $J = 13.6$, 5.4 Hz, 1H), 4.05 (s, 1H), 4.00 (d, J = 12.4 Hz, 1H), 3.77 (d, J = 12.4 Hz, 1H), 3.59 (d, $J = 13.6$ Hz, 1H), 3.31 (d, $J = 12.8$ Hz, 1H), 2.99 (dd, $J = 13.2$, 6.8 Hz, 1H), 2.71−2.67 (m, 1H), 2.65−2.61 (m, 1H), 2.48−2.43 (m, 2H), 1.80 (s, 3H); ¹³C NMR (126 MHz, DMSO- d_6 , 350 K) δ 173.4, 169.3, 168.7, 168.5, 166.1, 165.6, 165.4, 134.1, 133.8, 133.7, 133.2, 132.2, 130.4, 129.1, 128.8, 128.7, 127.7, 127.3, 126.7, 126.6, 126.4, 126.2, 125.9, 125.8, 125.7, 123.51, 123.46, 123.3, 122.2, 71.4, 67.5, 66.6, 62.0, 56.9, 56.7, 52.7, 51.1, 50.6, 44.4, 43.9, 42.5, 38.1, 36.0, 22.2; LC−MS analysis at 274 nm (C_{18} reversed-phase, 40 min, 5−100% H₂O/acetonitrile with 0.1% formic acid) $t_R = 20.3$ min; HRMS-ESI m/ z calcd for $C_{44}H_{41}N_5O_9$ $(M + H)^+$ 784.2977, found 784.2959.

(3S,5S,7′R,8a′S)-7′-(Acryloyloxy)-2′-(naphthalen-1-ylmethyl)- 1′,4′-dioxohexahydro-1′H-spiro[pyrrolidine-3,3′-pyrrolo[1,2-a] pyrazine]-5-carboxylic acid, dimer intermediate 24: HRMS-ESI m/z calcd for $C_{25}H_{25}N_3O_6 (M + H)^+$ 464.1816, found 464.1794.

(3S,5S,7′S,7″R,8′aS,8″aS)-1-Acetyl-2′,2″-bis[[4-[(benzyloxy) carbonyl]phenyl]methyl]-1′,1″,4′,4″-tetraoxo-7″-(pent-4-enoyloxy) dodecahydro-1″H-dispiro[pyrrolidine-3,3′:7′,3″-bis(pyrrolo[1,2-a] piperazine)]-5-carboxylic Acid, Trimer 29. Starting from amino acid 19 (279 mg, 0.85 mmol), the synthesis of trimer 29 followed the general trimer synthesis procedure with corresponding equivalence except changing the 1-naphthyl-pro4 amino acid 4 to benzyl carboxylate pro4 5 with an overall 33% isolated yield: ¹H NMR (500 MHz, CDCl₃, rt) δ 8.06–8.01 (m, 4H), 7.40–7.34 (m, 10H), 7.22−7.17 (m, 2H), 7.15 (t, J = 6.5 Hz, 2H), 5.82 (ddt, J = 16.6, 10.2, 6.3 Hz, 1H), 5.46 (dd, J = 9.0, 4.5 Hz, 1H), 5.36−5.31 (m, 6H), 5.29− 5.17 (m, 2H), 5.11−5.00 (m, 2H), 4.65−4.34 (m, 4H), 4.24 (dd, J = 25.9, 16.9 Hz, 1H), 4.08 (t, J = 10.8 Hz, 1H), 4.06−4.00 (m, 1H), 3.96−3.71 (m, 1H), 3.70−3.51 (m, 2H), 2.88 (dd, J = 12.9, 6.3 Hz, 1H), 2.66 (dd, J = 13.5, 6.5 Hz, 2H), 2.46 (q, J = 7.3 Hz, 2H), 2.40 (dd, J = 13.2, 6.3 Hz, 2H), 2.37–2.26 (m, 3H), 1.99 (s, 3H); ¹³C NMR (126 MHz, CDCl₃, rt) δ 172.9, 172.1, 168.2, 167.9, 165.8, 165.7, 165.0, 164.5, 164.2, 142.0, 141.4, 136.2, 135.9, 135.8, 130.6, 130.2, 129.9, 128.7, 128.6, 128.4, 128.3, 128.1, 126.2, 126.1, 116.0, 70.2, 69.2, 67.8, 67.1, 66.9, 66.8, 62.7, 58.0, 57.1, 56.3, 53.0, 52.0, 50.0, 49.3, 46.0, 43.6, 38.2, 36.3, 33.4, 28.7, 22.3; LC−MS analysis at 274 nm (C₁₈ reversed-phase, 40 min, 5-100% H₂O/acetonitrile with 0.1% formic acid) $t_R = 24.1$ min; HRMS-ESI m/z calcd for $C_{54}H_{53}N_5O_{13} (M + H)^+$ 980.3713, found 980.3708.

(3S,5S,7′S,7″R,8′aS,8″aS)-1-Acetyl-2′-[(4-bromophenyl)methyl]- 2″-[(4-methoxyphenyl)methyl]-1′,1″,4′,4″-tetraoxo-7″-(pent-4 enoyloxy)dodecahydro-1″H-dispiro[pyrrolidine-3,3′:7′,3″-bis- (pyrrolo[1,2-a]piperazine)]-5-carboxylic Acid, Trimer 30. Starting from amino acid 19 (509 mg, 1.56 mmol) and following the general synthesis procedure of trimer 26 except changing the pro4 amino acid 4 to the corresponding 6 and 7, trimer 30 (560 mg, 44% overall isolated yield) was obtained as a white solid: ¹H NMR (500 MHz, DMSO- d_6 , 350 K) δ 7.51 (d, J = 8.4 Hz, 2H), 7.21–7.12 (m, 2H), 7.10 (d, J = 8.7 Hz, 2H), 6.91−6.86 (m, 2H), 5.85 (ddt, J = 16.7, 10.3, 6.4

Hz, 1H), 5.36−5.27 (m, 1H), 5.07 (dtd, J = 17.2, 3.7, 1.6 Hz, 1H), 5.00 (dddd, $J = 6.7, 2.8, 2.2, 1.1$ Hz, 1H), 4.92 (d, $J = 16.3$ Hz, 1H), 4.78 (broad, 1H), 4.73 (dd, J = 15.1, 7.4 Hz, 1H), 4.60−4.53 (m, 1H), 4.50 (dd, J = 10.5, 6.8 Hz, 1H), 4.38 (d, J = 16.3 Hz, 1H), 4.31−4.09 (m, 2H), 4.02 (dd, J = 13.5, 5.3 Hz, 1H), 3.95−3.90 (m, 1H), 3.84 (d, J = 12.2 Hz, 1H), 3.74 (s, 3H), 3.46 (d, J = 13.5 Hz, 1H), 3.31−3.22 (m, 1H), 3.01−2.68 (m, 4H), 2.47−2.43 (m, 2H), 2.38−2.32 (m, 4H), 1.92−1.83 (m, 3H); ¹³C NMR (101 MHz, DMSO-d₆, rt, rotamers observed) δ 173.1, 172.33, 172.26, 169.4, 168.8, 168.7, 168.5, 168.4, 165.3, 164.6, 158.8, 158.6, 137.5, 137.3, 131.8, 131.3, 131.0, 130.0, 129.1, 128.7, 127.6, 120.6, 120.5, 120.4, 115.9, 114.3, 71.1, 67.4, 66.9, 66.5, 56.7, 56.1, 55.5, 55.4, 52.5, 51.4, 50.6, 49.9, 49.2, 46.6, 45.1, 44.6, 43.2, 37.4, 37.2, 35.7, 33.1, 28.7, 23.5, 22.5, 22.3, 21.9, 21.6, 19.4, 19.2, 19.0; HRMS-ESI m/z HRMS-ESI m/z calcd for $C_{39}H_{42}BrN_5O_{10}$ (M + H)+ 820.2188, found 820.2118.

(3R,5S,7′S,7″R,8′aS,8″aS)-1-Acetyl-2′-[(4-bromophenyl)methyl]- 2″-[(4-methoxyphenyl)methyl]-1′,1″,4′,4″-tetraoxo-7″-(pent-4 enoyloxy)dodecahydro-1"H-dispiro[pyrrolidine-3,3':7',3"-bis-(pyrrolo[1,2-a]piperazine)]-5-carboxylic Acid, Trimer 31. Starting from amino acid 19 (273 mg, 0.83 mmol) and following the general trimer synthesis procedure of 26 except changing the pro4 amino acid 4 to the corresponding 6 and 8, trimer 31 (260 mg, 38% overall isolated yield) was obtained as a white solid: $^1\mathrm{H}$ NMR (500 MHz, DMSO- d_6) δ 7.57–7.50 (m, 1H), 7.48–7.42 (m, 2H), 7.15 (d, J = 8.4 Hz, 2H), 7.09 (d, J = 8.6 Hz, 2H), 6.88 (d, J = 8.7 Hz, 2H), 5.85 (ddt, J = 16.7, 10.3, 6.3 Hz, 1H), 5.36−5.29 (m, 1H), 5.08 (ddd, J = 17.2, 3.4, 1.7 Hz, 1H), 5.01 (ddd, J = 10.3, 3.0, 1.4 Hz, 1H), 4.81−4.63 (m, 2H), 4.61−4.17 (m, 4H), 4.16−3.82 (m, 4H), 3.74 (s, 3H), 3.63−3.33 (m, 2H), 3.26−2.57 (m, 4H), 2.48−2.43 (m, 2H), 2.41−2.29 (m, 4H), 1.72 (s, 3H); ¹³C NMR (126 MHz, DMSO- d_6 , rt, rotamers observed) δ 173.0, 172.8, 172.3, 169.4, 169.0, 168.4, 165.5, 165.4, 164.4, 163.6, 158.8, 137.4, 131.5, 131.3, 130.1, 128.9, 128.7, 127.8, 120.3, 116.0, 114.5, 71.2, 68.8, 67.0, 58.1, 57.7, 56.8, 56.3, 56.1, 55.5, 52.6, 50.3, 50.0, 46.8, 46.0, 45.8, 44.7, 37.6, 37.4, 35.9, 35.0, 33.3, 28.8, 22.2, 21.5, 19.3; HRMS-ESI m/z calcd for $C_{39}H_{42}BrN_5O_{10} (M + H)^+$ 820.2188, found 820.2208.

(3R,5S,7 ′ S,7 ″ R,8 ′ aS,8 ″ aS)-1-Acetyl-5-[[(1S,2S)-2 aminocyclohexyl]carbamoyl]-2″-(2-methylpropyl)-2′-[(4 nitrophenyl)methyl]-1′,1″,4′,4″-tetraoxododecahydro-1″H-dispiro- [pyrrolidine-3,3′:7′,3″-bis(pyrrolo[1,2-a]piperazine)]-7″-yl Pent-4 enoate, 33. Starting from amino acid 19 (557 mg, 1.7 mmol) and following the general trimer synthesis procedure of 26 except changing the alkylated pro4 4 to the corresponding 9 and 10, crude trimer 32 (780 mg, theoretically 1.08 mmol) was obtained as a brown solid, which was used in the next step without further purification. In a flame-dried round-bottom flask, crude trimer 32 (780 mg, theoretically 1.08 mmol, 1.0 equiv) and HATU (616 mg, 1.62 mmol, 1.5 equiv) were dissolved in 20 mL of dry 20 DMF/DCM (1:1) at room temperature, followed by the addition of DIPEA (561 μ L, 3.24 mmol, 3.0 equiv) in one portion. The reaction mixture was stirred for 10 min and then slowly added to another flame-dried round-bottom flask containing diaminocyclohexane (740 mg, 6.48 mmol, 6.0 equiv) in 15 mL of DMF within 2 h. The reaction was stirred overnight at room temperature until completion as indicated by LC−MS. The reaction mixture was then quenched by addition of satd NH_4Cl (aq) (50 mL), and the product was extracted with EtOAc (100 mL \times 3). The combined organic portion was washed sequentially with 2×20 mL brine and dried over anhydrous $Na₂SO₄$. After filtration and removal of solvent under vacuum, the crude residue was purified by reversedphase C18 column using a 5−95% H2O/acetonitrile gradient as eluting solvent to give the pure product 33 as a yellow colored powder (270 mg, 0.33 mmol, 19% overall isolated yield from 19): ¹H NMR (500 MHz, DMSO- d_6) δ 8.19 (d, J = 8.7 Hz, 2H), 7.92–7.78 (m, 1H), 7.61−7.42 (m, 2H), 5.84 (ddt, J = 16.7, 10.3, 6.4 Hz, 1H), 5.29 (t, J = 4.7 Hz, 1H), 5.06 (ddd, J = 17.2, 3.3, 1.6 Hz, 1H), 4.99 (ddd, J = 10.3, 3.0, 1.3 Hz, 1H), 4.81−4.68 (m, 1H), 4.64 (dd, J = 10.8, 6.6 Hz, 1H), 4.49 (s, 1H), 4.37–4.18 (m, 1H), 4.10 (d, J = 12.1 Hz, 1H), 4.01 (d, J $= 12.2$ Hz, 1H), 3.97 (dd, J = 13.5, 5.3 Hz, 1H), 3.53 (dd, J = 14.3, 8.2 Hz, 1H), 3.41 (d, $J = 13.5$ Hz, 1H), 3.01 (dd, $J = 14.2$, 6.7 Hz, 1H), 2.89 (s, 2H), 2.82 (dd, J = 13.0, 6.8 Hz, 1H), 2.46−2.41 (m, 2H),

2.36−2.30 (m, 3H), 2.26−2.22 (m, 1H), 2.14−2.02 (m, 1H), 2.00− 1.88 (m, 3H), 1.80−1.74 (m, 1H), 1.73−1.65 (m, 2H), 1.38−1.30 (m, 1H), 1.26−1.19 (m, 2H), 0.86 (dd, J = 22.1, 6.7 Hz, 6H); 13C NMR (126 MHz, DMSO- d_6 , 350 K, from HSQC and HMBC, overlap observed) δ 171.3, 171.2, 170.8, 168.4, 168.0, 167.9, 167.3, 164.7, 164.4, 145.3, 136.7, 136.3, 136.1, 136.0, 126.6, 122.6, 114.6, 114.5, 114.3, 70.3, 70.1, 69.9, 69.8, 69.4, 65.6, 65.5, 57.3, 55.7, 55.6, 55.4, 55.2, 55.0, 54.5, 52.4, 50.7, 50.5, 50.0, 48.3, 47.4, 44.2, 44.1, 36.2, 36.1, 34.5, 34.3, 31.7, 29.3, 29.1, 28.4, 27.3, 26.9, 22.3, 20.8, 18.7; HRMS-ESI m/z calcd for $C_{41}H_{54}N_8O_{10} (M + H)^+$ 819.4036, found 819.4030.

General Synthetic Procedure of Macrocycles 38−40. A representative synthesis of macrocycle 38 is described as follows: In a flame-dried round-bottom flask, trimer 26 (41 mg, 0.05 mmol, 1.0 equiv) and HATU (23 mg, 0.06 mmol, 1.2 equiv) were dissolved in dry 1.65 mL DMF/DCM (10:1) at room temperature followed by the addition of DIPEA (21 μ L, 0.12 mmol, 2.4 equiv) in one portion. The reaction mixture was stirred for 10 min and then slowly added to another flame-dried round bottle containing diaminocyclohexane (2.57 mg, 0.023 mmol, 0.45 equiv) in 0.5 mL of DMF (this solution was prepared from a stock solution of 25.7 mg diaminocyclohexane in 5 mL DMF) within 2 h. The reaction was stirred overnight at room temperature until completion as indicated by LC−MS. The reaction mixture was then quenched by addition of satd $NH₄Cl$ (aq) (10 mL), and the product was extracted with EtOAc (50 mL). The combined organic portion was washed sequentially with $3\times$ satd NH₄Cl (aq) and $2\times$ brine and dried over anhydrous Na₂SO₄. After filtration and removal of solvent under vacuum, the crude residue was purified by a reversed-phase C_{18} column using 5−95% H₂O/acetonitrile gradient as eluting solvent to give pure product 34 as white powder (30 mg, 0.017 mmol, 78% isolated yield).

Compound 34 (17 mg, 0.01 mmol, 1.0 equiv) in 4 mL of DCM was degassed using bubbling argon for 5 min, followed by the addition of Hoveyda−Grubbs Catalyst II (3.1 mg, 5 × 10[−]³ mmol, 0.5 equiv) in one portion. The reaction was stirred under inert atmosphere for 24 h until completion as indicated by LC−MS. The solvent was removed under vacuum, and the residue was purified through reversed-phase C_{18} column using 5−95% H₂O/acetonitrile gradient as eluting solvent to give pure macrocycle 38 as a white powder (12 mg, 7.2×10^{-3} mmol, 72% isolated yield).

(3S,5S,7′S,7″R,8′aS,8′aS,8″aS)-5-[[(1S,2S)-2-[[(3S,5S,7′S,7″R,8′aS,- 8″aS)-1-acetyl-2′,2″-bis(naphthalene-1-ylmethyl)-1′,1″,4′,4″-tetraoxo-7″-(pent-4-enoyloxy)dodecahydro-1″H-dispiro[pyrrolidine-3,3′:7′,3″bis(bis(pyrrolo[1,2-1]piperazine)]-5-yl]amido]cyclohexyl] carbamoyl]-1-acetyl-2″,2″-bis(naphthalene-1-ylmethyl)-1′,1″,4′,4″ tetraoxododecahydro-1″H-dispiro[pyrrolidine-3,3′:7′,3″-bis(pyrrolo [1,2-a]piperazine)]-7"-yl pent-4-enoate, 34: 1 H NMR (500 MHz, DMSO- d_6 , 350 K) δ 8.11 (t, J = 4.1 Hz, 3H), 8.03 (s, 2H), 7.97 (d, J = 9.3 Hz, 2H), 7.95 (d, J = 6.6 Hz, 2H), 7.86−7.80 (m, 4H), 7.63−7.48 (m, 9H), 7.47−7.41 (m, 4H), 7.24−7.04 (m, 4H), 5.83 (qd, J = 12.1, 6.1 Hz, 2H), 5.50 (d, J = 17.4 Hz, 2H), 5.46–5.34 (m, 3H), 5.29 (s, 1H), 5.06 (d, J = 17.2 Hz, 1H), 5.00 (d, J = 9.5 Hz, 1H), 4.97−4.66 (m, 6H), 4.54 (broad s, 2H), 4.22 (broad s, 1H), 4.03 (broad s, 1H), 3.99 (d, J = 12.5 Hz, 2H), 3.78 (d, J = 12.5 Hz, 2H), 3.66−3.30 (m, 4H), 2.98 (dd, J = 13.1, 6.7 Hz, 2H), 2.83−2.63 (m, 4H), 2.47−2.07 (m, 16H), 1.80−1.44 (m, 12H), 1.14 (broad s, 6H); 13C NMR (126 MHz, DMSO- d_6) δ 174.4, 172.1, 170.7, 168.6, 168.3, 165.6, 162.8, 137.4, 133.8, 133.2, 133.0, 130.5, 130.3, 129.1, 127.8, 126.7, 126.4, 125.9, 123.3, 123.0, 122.8, 122.2, 119.0, 116.7, 115.8, 114.4, 112.2, 70.9, 67.5, 56.9, 56.6, 52.6, 50.7, 44.0, 38.3, 36.1, 33.4, 32.1, 31.6, 29.4, 29.2, 29.0, 28.9, 28.7, 27.4, 26.2, 24.6, 22.4, 14.1; LC−MS analysis at 274 nm (C_{18} reversed-phase, 40 min, 5–100% H₂O/acetonitrile with 0.1% formic acid) $t_R = 26.5$ min; HRMS-ESI m/z calcd for $C_{98}H_{100}N_{12}O_{16}$ (M + 2H)/2⁺ 851.8779, found 851.8763.

Macrocycle **38:** ¹H NMR (500 MHz, CDCl₃, rt) δ 8.05 (d, J = 8.3 Hz, 2H), 7.99−7.86 (m, 6H), 7.80 (t, J = 7.5 Hz, 4H), 7.56 (broad s, 8H), 7.41 (broad s, 4H), 7.02 (d, J = 6.6 Hz, 2H), 6.95 (d, J = 6.6 Hz, 2H), 6.85 (s, 2H), 5.69 (dd, J = 23.7, 17.1 Hz, 4H), 5.51 (d, J = 24.6 Hz, 4H), 4.94 (d, $J = 17.3$ Hz, 2H), 4.76 (d, $J = 16.7$ Hz, 2H), 4.62 (dd, J = 16.5, 10.2 Hz, 4H), 4.46 (t, J = 8.4 Hz, 2H), 4.32–4.05 (m, 4H), 3.93 (d, J = 13.2 Hz, 2H), 3.84 (d, J = 11.4 Hz, 2H), 3.66 (d, J =

13.9 Hz, 2H), 3.52 (s, 2H), 3.40 (d, J = 11.4 Hz, 2H), 3.07−2.77 (m, 3H), 2.78−2.62 (m, 3H), 2.62−2.53 (m, 2H), 2.53−2.21 (m, 10H), 1.73−1.56 (m, 12H), 1.21−1.05 (m, 4H); 13C NMR (126 MHz, CDCl3, rt) δ 172.1, 170.2, 169.8, 168.1, 167.6, 165.7, 165.4, 133.9, 133.8, 131.8, 130.6, 130.2, 130.1, 129.3, 128.9, 128.5, 128.2, 126.7, 126.5, 126.3, 126.2, 125.4, 125.2, 122.7, 122.0, 121.7, 121.5, 70.2, 67.9, 67.4, 57.1, 56.4, 54.7, 53.5, 53.2, 52.0, 50.6, 44.2, 44.0, 38.5, 36.6, 35.5, 33.8, 31.7, 29.7, 27.5, 24.2, 22.3; LC−MS analysis at 274 nm (C₁₈ reversed-phase, 30 min, 5−95% H2O/acetonitrile with 0.1% formic acid) $t_R = 25.5$ min; HRMS-ESI m/z calcd for $C_{96}H_{96}N_{12}O_{16}$ (M + 2H)/2⁺ 837.8622, found 837.8607.

Macrocycle 39. Starting from trimer 26 (82 mg, 0.1 mmol, 1.0 equiv), macrocycle 39 was obtained through intermediate 35 by following the identical procedure as macrocycle 38 except changing diaminocyclohexane to 1,2-ethanediamine (2.7 mg, 0.045 mmol, 0.45 equiv) (49% overall isolated yield, 67% in the first step and 73% in the second step): LC−MS analysis at 274 nm (C₁₈ reversed-phase, 30 min, 5−95% H₂O/acetonitrile with 0.1% formic acid) t_R = 24.9 min; HRMS-ESI m/z calcd for $C_{92}H_{90}N_{12}O_{16}$ $(M + H)^+$ 1619.660, found 1619.664.

Macrocycle 40. Starting from trimer 27 (110 mg, 0.138 mmol, 1.0 equiv), macrocycle 40 was obtained through intermediate 36 by following the identical procedure as macrocycle 38 except changing diaminocyclohexane to 1,2-ethanediamine (3.7 mg, 0.062 mmol, 0.45 equiv) (36% overall isolated yield, 60% in the first step and 61% in the second step): LC−MS analysis at 274 nm (C_{18} reversed-phase, 30 min, 5−95% H₂O/acetonitrile with 0.1% formic acid) $t_R = 23.5$ min; HRMS-ESI m/z calcd for $C_{90}H_{86}N_{12}O_{16}$ $(M + 2H)/2^+$ 796.8231, found 796.8251. Intermediate 36: HRMS-ESI m/z calcd for $C_{92}H_{90}N_{12}O_{16}$ $(M + 2H)/2$ ⁺ 810.3372, found 810.3347.

(3S,5S,7′S,7″R,8′aS,8″aS)-5-[[2-[(3S,5S,7′S,7″R,8′aS,8″aS)-1-Acetyl-2′,2″-bis(naphthalen-1-ylmethyl)-1′,1″,4′,4″-tetraoxo-7″-(prop-2-enoyloxy)dodecahydro-1″H-dispiro[pyrrolidine-3,3′:7′,3″-bis- (pyrrolo[1,2-a]piperazine)]-5-ylformamido]ethyl]carbamoyl)-1-acetyl-2′,2″-bis(naphthalen-1-ylmethyl)-1′,1″,4′,4″-tetraoxododecahydro-1″H-dispiro[pyrrolidine3,3′:7′,3″bis(pyrrolo[1,2-a]piperazine)]- 7"-yl Prop-2-enoate, Compound 37. Starting from trimer 28 (20 mg, 0.025 mmol, 1.0 equiv) and 1, 2-ethanediamine (0.68 mg, 0.011 mmol, 0.45 equiv), compound 37 (14 mg, 8.8 × 10[−]³ mmol, 80% yield) was obtained by following the same general procedure as for 35: HRMS-ESI m/z calcd for $C_{90}H_{86}N_{12}O_{16}$ $(M + 2H)/2$ 796.8231, found 796.8254. The metathesis reaction of compound 37 resulted in a HPLC trace that could not be interpreted and no mass corresponding to the product was found.

Synthetic Procedure of Macrocycle 46. At room temperature, trimer 30 (60 mg, 0.073 mmol, 1.0 equiv) and HATU (56 mg, 0.147 mmol, 2.0 equiv) were dissolved in 1.5 mL of anhydrous DMF/DCM $(2:1)$ in a flame-dried vial, followed by the addition of DIPEA (51 μ L, 0.293 mmol, 4.0 equiv) in one portion. The reaction mixture was stirred for 10 min and transferred dropwise to another flame-dried vial containing amine 33 (60 mg, 0.073 mmol, 1.0 equiv) in 1 mL of anhydrous DMF under inert atmosphere. The reaction mixture was stirred overnight at room temperature until completion as indicated by LC−MS. The reaction mixture was then quenched by addition of satd NH4Cl (aq) (10 mL), and the products were extracted with EtOAc (50 mL). The combined organic portion was washed sequentially washed with $3\times$ satd NH₄Cl (aq), $3\times$ satd NaHCO₃ (aq), and $2\times$ brine and dried over anhydrous $Na₂SO₄$. After filtration and removal of solvent under vacuum, the crude residue was purified by reversedphase C18 column using 5−95% H2O/acetonitrile gradient as eluting solvent to give pure dialkene as white powder 42 (70 mg, 0.043 mmol, 59% isolated yield). The dialkene intermediate 42 was then subjected to the metathesis reaction as described in synthesis of macrocycle 38 to give macrocycle 44 with an 80% isolated yield. The hydrogenation of 44 with palladium on activated carbon with H2 balloon afforded pure macrocycle 46 as a white powder with an 80% isolated yield: $^1\mathrm{H}$ NMR (500 MHz, C_6D_6 , rt, TFA-d added) δ 7.99 (d, J = 8.2 Hz, 2H), 7.47−7.41 (m, 1H), 7.38 (d, J = 8.4 Hz, 2H), 7.36−7.29 (m, 1H), 7.04 $(d, J = 8.7 \text{ Hz}, 2H), 6.81 (d, J = 8.7 \text{ Hz}, 2H), 5.09-4.78 (m, 3H),$ 4.76−4.65 (m, 3H), 4.61 (d, J = 15.5 Hz, 1H), 4.57−4.28 (m, 7H),

4.24 (d, J = 12.5 Hz, 1H), 4.11 (d, J = 12.5 Hz, 1H), 4.06−3.97 (m, 2H), 3.96−3.89 (m, 1H), 3.85−3.78 (m, 1H), 3.76−3.70 (m, 2H), 3.69−3.63 (m, 1H), 3.57−3.40 (m, 4H), 3.30 (s, 3H), 3.16−2.96 (m, 2H), 2.83 (dd, J = 13.9, 6.3 Hz, 1H), 2.75−2.67 (m, 1H), 2.64−2.50 (m, 2H), 2.49−2.41 (m, 2H), 2.41−2.30 (m, 2H), 2.24−2.10 (m, 4H), 1.96 (d, J = 9.9 Hz, 4H), 1.93−1.87 (m, 2H), 1.86−1.77 (m, 2H), 1.72−1.65 (m, 1H), 1.55−1.43 (m, 5H), 1.37−1.26 (m, 8H), 1.19− 1.07 (m, 8H), 0.96-0.88 (m, 4H); ¹³C NMR (126 MHz, C_6D_6 , rt, collected from HSQC and HMBC, overlapping observed) δ 171.4, 169.7, 169.3, 159.4, 147.6, 142.2, 132.5, 127.9, 127.8, 126.9, 126.7, 124.2, 114.8, 70.9, 67.4, 67.0, 57.5, 57.4, 56.9, 56.8, 56.0, 55.9, 54.7, 53.4, 53.2, 51.1, 50.6, 50.1, 49.5, 46.2, 46.0, 45.8, 36.3, 35.7, 35.5, 35.2, 35.1, 34.9, 33.4, 31.3, 31.2, 29.7, 28.7, 19.1, 19.0, 18.7, 13.9; NMR including COSY, HSQC, HMBC, and ROESY, and characterization are included in the Supporting Information; LC−MS analysis at 274 nm (C₁₈ reversed-phase, 40 min, 5–100% H₂O/acetonitrile with 0.1% formic acid) $t_R = 22.2$ min; HRMS-ESI m/z calcd for $C_{78}H_{92}BrN_{13}O_{19}$ $(M + H)^+$ 1594.58[89, found 1594.5886.](http://pubs.acs.org/doi/suppl/10.1021/acs.joc.5b01109/suppl_file/jo5b01109_si_001.pdf)

Macrocycle 47. Starting from trimer 33, macrocycle 47 was obtained with a 20% overall isolated by following the synthesis of macrocycle 46 except changing trimer 30 to 31: ¹H NMR (500 MHz, C_6D_6 , rt, TFA-d added, several conformations observed) δ 7.45 (dd, J = 18.8, 8.2 Hz, 2H), 7.34 (d, J = 7.9 Hz, 2H), 7.27−7.20 (m, 1H), 7.13−7.08 (m, 2H), 7.04−6.99 (m, 2H), 6.82 (dd, J = 22.9, 8.2 Hz, 2H), 6.76−6.67 (m, 3H), 4.90−4.67 (m, 6H), 4.66−4.59 (m, 1H), 4.55−4.46 (m, 2H), 4.44−4.34 (m, 2H), 4.32−4.26 (m, 1H), 4.26− 4.13 (m, 3H), 4.12−4.03 (m, 2H), 4.03−3.86 (m, 3H), 3.86−3.76 (m, 2H), 3.75−3.60 (m, 3H), 3.59−3.46 (m, 3H), 3.43 (d, J = 13.8 Hz, 1H), 3.28−3.16 (m, 2H), 3.15−3.06 (m, 3H), 3.06−2.61 (m, 6H), 2.55−2.37 (m, 4H), 2.16−2.08 (m, 4H), 1.99−1.90 (m, 3H), 1.88 (s, 3H), 1.82−1.74 (m, 4H), 1.60−1.50 (m, 4H), 1.07 (m, 16H, overlap with grease and impurities, not identified); ¹³C NMR (126 MHz, C_6D_6 , rt, collected from HSQC and HMBC, overlapping observed) δ 181.9, 179.2, 176.1, 174.3, 173.5, 170.3, 166.2, 165.5, 159.6, 150.0, 149.4, 148.3, 147.8, 142.6, 139.2, 133.4, 131.3, 130.7, 130.0, 127.8, 124.7, 123.1, 87.6, 74.3, 72.3, 67.9, 67.6, 67.4, 64.6, 53.9, 50.6, 47.6, 46.8, 46.1, 35.5, 35.2, 34.9, 33.9, 32.6, 31.7, 31.5, 30.5, 29.8, 29.3, 29.0, 28.0, 27.2, 24.8, 23.3, 21.8, 19.7, 19.5, 14.5; NMR including COSY, HSQC, HMBC and ROESY, and characterization are included in the Supporting Information; LC−MS analysis at 274 nm (C₁₈ reversedphase, 40 min, 5–100% H₂O/acetonitrile with 0.1% formic acid) t_R = 21.9 min; HRMS-ESI m/z calcd for $C_{78}H_{92}BrN_{13}O_{19}$ $(M + H)^+$ [1594.5889, found 1594.5](http://pubs.acs.org/doi/suppl/10.1021/acs.joc.5b01109/suppl_file/jo5b01109_si_001.pdf)915.

■ ASSOCIATED CONTENT

S Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.joc.5b01109.

General information, ${}^{1}H$ and ${}^{13}C$ NMR spectra, and [LCMS spectra \(PD](http://pubs.acs.org)F)

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Notes

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